# M.Sc. 4th Semester Examination, 2021 <br> CHEMISTRY <br> (Organic Chemistry Special Practical) <br> Paper : CHEM 405E(PR) <br> Course Id : 41465 

## Time: 2 hours

Full Marks: 40
The figures in the margin indicate full marks
Candidates are required to give their answers in their own words as far as practicable
Answer any four of the following questions:
$10 \times 4=40$

1. How will you synthesize coumarin in the laboratory? Name the chemicals and apparatus required for its synthesis. Give a plausible mechanism of the reaction involved.

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3+(2+2)+3=10
$$

2. Describe the laboratory method for the synthesis of $\beta$-hydroxynaphthaldehyde. Write the name of the reaction. Write down the chemicals and apparatus required for its synthesis. Propose a plausible mechanism of the reaction involved. Draw the structure of electron deficient reaction intermediate formed in this reaction. $2+1+(2+2)+2+1=10$
3. Write the method for the preparation 1,4-ditertiarybutyl benzene involving Friedel-Crafts reaction. Why is anhydrous aluminium chloride used? Mention the chemicals and apparatus required for its synthesis. Give a plausible mechanism of the reaction involved. Does phenol undergo Friedel-Crafts reaction?

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2+1+(2+2)+2+1=10
$$

4. Describe the laboratory method for the synthesis of p-acetamidobenzene sulphonyl chloride. Why dry acetanilide is used? Name the chemicals and apparatus required for its synthesis. Draw a plausible mechanism involved in the reaction. $2+1+(2+2)+3=10$
5. How will you synthesize benzanilide from benzophenone oxime in the laboratory? Write the name of the reaction. Mention the chemicals and apparatus required for its synthesis. Give a probable mechanism of the reaction.

$$
2+1+(2+2)+3=10
$$

6. Describe the synthesis of phenylazo- $\beta$-naphthol in the laboratory. Write the name of the reaction. Name the chemicals and apparatus required for its synthesis. Propose a viable mechanism of the reaction involved. Why is reaction temperature maintained at $0-5{ }^{\circ} \mathrm{C}$ ?

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2+1+(2+2)+2+1=10
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